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The following reaction has an equilibrium constant of 620 at a certain temperature. Calculate the equilibrium concentrations of all species if 4.5 mol of each component were added to a 3.0 L flask.  $\text{H}_2(\text{g}) + \text{F}_2(\text{g}) \rightleftharpoons 2 \text{HF}(\text{g})$  Determine molarity of solutions  $[4.5 \text{ mol} / 3.0\text{L}] = 1.5 \text{ M}$  of all 3 solutions

~~Equilibrium Practice Problems~~

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32: Molarity

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14) Gas X<sub>2</sub> reacts with gas Y<sub>2</sub> to give gaseous XY according to the equation X<sub>2</sub> + Y<sub>2</sub>  $\leftrightarrow$  2XY. K<sub>eq</sub> = 2.6x10<sup>-3</sup>. What is K<sub>eq</sub> for the reverse reaction? a) 6.8x10<sup>-6</sup> b) 5.1x10<sup>-2</sup> c) 2.0x10<sup>1</sup> d) 3.8x10<sup>2</sup> 15) Which one of the following statements about a reaction indicates that the products are favored at equilibrium?

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Workforce transformation based around new models of care and "skill-mix" change in the form of 5000 new "non-medical roles" to operate alongside GPs is an aspirational solution,<sup>1</sup> though generating the right balance of GPs/non-GPs is not without controversy.<sup>2</sup> Although practice nurses have been working in extended roles in general practice for a long time<sup>3</sup> there are other "new" roles ...

~~Skill-mix change and the general practice workforce ...~~

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Equilibrium is a skill located in the Mages Guild (which can be found in the Guild skill tree). Equilibrium is a base skill, and can be morphed into Spell Symmetry or Balance.

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